Electronic and geometric structure of the h-BN/Rh(111) Nanomesh

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within an EU-projoct lead by J.Osterwalder (Univ.Zürich) and a joint DFG/FWF project H.Over/P.Blaha





h-BN on transition metal-surfaces



thermal decomposition of borazine (HBNH)₃
 on hot TM metal surfaces (~1000K)



borazine (HBNH)₃

forms simple (1x1) or very complex structures as seen by STM

h-BN/Ni(111); Cu(111)



h-BN/Rh(111); Ru(001); Pt(111)









a_o [Å]





Only N on "top" of Ni gives stable structures

(top,fcc) by only 9 meV/BN more stable than (top,hcp)







$$I_t \propto \sum_{v} \left| \Psi_v(r_0) \right|^2 \delta(E_v - E_F)$$

Exp: three different sites are visible. Which site is dark? Which white ?



Electron densities at E_F +/- 0.08eV r₀: 2.5A from surface



6.00e-05 5.90e-05 5.80e-05 5.70e-05 5.60e-05 5.50e-05 5.40e-05 5.20e-05 5.20e-05 5.10e-05 5.00e-05



1.08e-05 1.05e-05 1.02e-05 9.90e-06 9.60e-06 9.30e-06 9.00e-06



 $\Delta \Phi$

15

10

(Bader's AIM method):

R Ni Ν free h-BN-l: -0.56 +0.56 e⁻ h-BN/Ni: -0.59 + 0.65 - 0.06

FIG. 2. He I normal emission UP-spectra for Ni(111) and h-BN/Ni(111). The work function decrease and the shift of the Λ_{-} Ni d band feature upon formation of the the k-BN layer indicate a decrease of the Ni magnetic moment (for details see text).

10 Binding energy (eV)

electrostatic picture with 1A charge separation \rightarrow 2eV shift

additional 0.06e should fill Ni-dn states: -> smaller moment













- formation driven by Rh-BN lattice mismatch (7%)
 - → 12x12 Rh / 13x13 BN superlattice
- two incomplete layers of h-BN
- holes in h-BN can dock molecules or clusters

S.Berner etal., Angew.Chemie





- high energy costs due to broken BN bounds !
- no reason for second layer to be incomplete !
- STM measured corrugation is small compared to distance between layers in h-BN !
- interaction of h-BN with Rh(111) ?
- why is h-BN/Rh(111) so different from h-BN/Ni(111) and does not form a flat overlayer ?
- propose an alternative model, that explains observed STM images and UPS σ band splitting ?



single layer model of the h-BN/Rh(111) nanomesh



single continuous BN layer: 13x13 BN on top of 12x12 Rh(111)



DFT simulation:

quite heavy: 3 layers TM + BN: 1108 atoms/cell (metal!)

WIEN2k (all electron APW+lo full-potential calculations)

- Hamiltonian size ~50000-70000 (50-100GB memory)
- 64-512 cores (Xeons), 120-20min per SCF iteration, 30 iterations/scf
- full structural optimization

corrugated h-BN layer on Rh(111)



2.7

2.6

2.5

2.4

2.3

2.2

B-high

B-low

Rh



"high": "top" empty

R.Laskowski, P.Blaha, Th.Gallauner, K.Schwarz, Single layer model of the h-BN nanomesh on the Rh(111) surface Phys.Rev.Lett. 98, 106802 (2007)





distance of N from metal layer (shown in 3x3 supercell)



- h-BN layer is corrugated, amplitude ~ 0.5 (Pt) -1.6 (Ru,Rh) Å
- flat region of BN close to metal
- surrounding rims are made of two maxima with slightly different height



Ab initio study of h-BN nanomeshes on Ru(001), Rh(111), and Pt(111) R.Laskowski, P.Blaha, PRB 81, 075418 (2010)







Comparing STM images (Tersoff-Hamann)





-2V: Goriachko et al., Langmuir **2007,** *23,* 2928

+0.07V: Preobrajenski et al. PRB 75, 245412









N 1s -XPS core level shifts









B-K edge in BN and BN/Ni(111)

Preobrajenski etal, PRB70, 165404 (2004): "The experiments contradict recent DFT calculations by Grad etal."













Subsurface Cation Vacancy Stabilization of the Magnetite (001) Surface

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R. Bliem, E. McDermott, P. Ferstl, M. Setvin, O. Gamba, J. Pavelec, M.A. Schneider, M. Schmid, U. Diebold, P. Blaha, L. Hammer, G. S. Parkinson, **Science** 346, 1215 (2014) R.Bliem et.al., **PRB** 92, 075440 (2015)







Magnetite is a natural mineral (lodestone)

- it is a permanent magnet (T_C=858 K) and was used as first compass in ancient time
- technological important material
 - magnetic recording
 - biomagnetism (magnetoreception)
 - catalyst for ammonia synthesis
 - half-metallic ferromagnet (spintronics)
 - Corrosion
- Cubic spinel structure at room temperature
- "Verwey" transition at ~120 K





Magnetite Fe₃O₄



Fe₃O₄, magnetite

phase transition between a mixed-valence and a charge-ordered configuration



The accompanying graph shows $\log \rho$ against 1/T for two hars: I with FcO: Fe₂O₃ = 1:1.025, and II with FeO: Fe₂O₃ = 1:1.08. All details of the curves are in full accordance with the picture proposed above for the nature of the transition and our concept of the eation arrangement in the Fe₃O₄ (and the γ -Fe₂O₃) lattice. In further support of our views, we found that sample I shows a distinct drop in the susceptibility for weak magnetic fields at about 117° K., whereas with sample II the corresponding effect is much weaker.

 $2 \operatorname{Fe}^{2.5+} \rightarrow \operatorname{Fe}^{2+} + \operatorname{Fe}^{3+}$

cubic inverse spinel structure AB₂O₄





 \rightarrow small, but complicated coupling between lattice and charge order





stimulated by U. Diebold (Z.Novotny et al., PRL 108, 216103 (2012))

- Fe_{oct}^{2.5+}O₂ planes, forming "rows" along [110]
- Fe_{tet}³⁺ planes
- Fe_{oct}^{2.5+}O₂ planes, forming "rows" along [1-10]

—

surface Fe_{oct} atoms are all
 Fe³⁺ to compensate (partly)
 the charged surface layer









An Augmented Plane Wave Plus Local Orbital Program for Calculating Crystal Properties

Peter Blaha et al. (Vienna University of Technology) http://www.wien2k.at

- surface slab calculations using (√2×√2)-17 and (2×2)-17 layer models (up to 248 atoms/cell)
- GGA+U (Fe-3d: U=3.8 eV)
- Rkmax=7, 3x3x1 k-points





The Fe₃O₄(001) Surface



Fe₃O₄(001) exhibits ($\sqrt{2}x\sqrt{2}$)R45° superstructure due to small lateral relaxations of surface atoms (≈ 0.1 Å)

In STM, we see the Fe_{oct} atoms



(4.38 x 6.43) nm² 1 V, 0.1 nA

Structure determined by DFT and LEED-IV:



found by R. Pentcheva *et al.*, PRL **94**, 126101 (2005) **explained** by DFT+U calc.: Lodziana, PRL 99, 206402 (2007): charge and orbital order in **sub-surface** Fe_{oct}O₂-rows







Two obvious sites for **ad-atom adsorption**: Bulk continuation Fe_{tet} sites "W" and "N"

In STM, we see the Fe_{oct} atoms



(4.38 x 6.43) nm² 1 V, 0.1 nA









experiments in U.Diebolds group:

Z.Novotny et al., PRL 108, 216103 (2012)

single Au atoms stable up to 400 C

Au adsorbs exclusively on narrow site



The Fe₃O₄(001) Surface



In STM, Au, Ag, Pt, Pd adsorb only at the N site.





DFT does not find any difference between Meadsorption on the W or N site G.Parkinson et al., Nature Mat. 12, 724 (2013) something must block this site !!!



Surface reconstruction including non-stoichiometry ?

 Pentcheva (and we also) have tried several different O-vacancy structures
 non of them are stable.



Fe-oxides contain mostly Fe-vacancies, not Ovacancies

γ-Fe₂O₃: Maghemite is a "Magnetite" with 1/6 Fe_{oct} vacancies

Subsurface cation vacancy structure

Fe₃O₄(001) surface reconstruction



Fe_{tet}-interstitial + **Distorted bulk truncation** sub-subsurface Fe_{oct} **B** Subsurface cation vacancy structure Fe_{oct} Feint Fe_{tet} **Fe**vac

new model:

much stronger Fe_{oct} corrugation



-3.0

-2.5

-2.0

Thermodynamic stability

WIEN



-^{1.5} μ_o (eV)

-1.0

-0.5

0.0





- Pentcheva et al. 2008: Refinement of the distorted bulkterminated structure with R=0.34
- new exp. by L.Hammer: new model refines to R=0.12
 old exp. can be refined with new model to the same R=0.12



refined exp. positions and theor. positions agree within 0.05 Å (all within the exp. error)





Fe_{int} blocks selectively adatom adsorption on this site

adsorbtion site



blocked site But: the **blocked** site is the **N** site (in contrast to exp.) !!??

STM measurements and simulations



Iarge corrugation at low bias voltage (narrow site at x)





STM measurements and simulations



at intermediate bias the rows "straighten"





STM measurements and simulations



at large bias the previous "narrow" site looks "wide"















STM with different bias voltage





- "N" site is actually the "wide" site for low bias
- "N" and "W" change in STM with the bias voltage
- STM probes electron densities, not "atomic positions"







• The new SCV-model of the Fe₃O₄ (001) surface explains

- LEED-IV measurements
- fixes polar catastrophe
- explains STM images (bias dependency)
- explains site-selective TM-atom adsorption
- spintronics problems



allows investigations of "single-atom" catalysis (CO oxidation with Pt)





- U. Diebold + G. Parkinson + group at IAP (TU Vienna)
- L. Hammer (Univ. Erlangen)
- E. McDermott + R. Bliem (calculations)

R. Bliem, E. McDermott, P. Ferstl, M. Setvin, O. Gamba, J. Pavelec, M.A. Schneider, M. Schmid, U. Diebold, P. Blaha, L. Hammer, G. S. Parkinson, Science 346, 1215 (2014)



Thank you for your attention !